BROMINE PENTAFLUORIDE

CAUTIONARY RESPONSE INFORMATION 4. FIRE HAZARDS 4.1 Flash Point: Common Synonyms Liquefied das Colorless Irritating odor Not flammable. Chemical is strong oxidizer and may cause fire in contact with organic materials such as wood, Reacts violently with water. Poisonous vapor is produced cotton, or straw. 4.2 Flammable Limits in Air: Not flammable Restrict access 4.3 Fire Extinguishing Agents: Not pertinent Avoid contact with liquid and vapor. Notify local health and pollution control agencies. 4.4 Fire Extinguishing Agents Not to Be Used: Do not use water or foam on adiacent fires. Protect water intak Special Hazards of Combustion Products: Toxic and irritating fumes of Not flammable Fire May cause fire on contact with combustibles hydrogen fluoride and bromine may form in fires. Will increase the intensity of a fire. POISONOUS GASES MAY BE PRODUCED IN FIRE. Containers may explode in fire. 4.6 Behavior in Fire: Containers may burst when exposed to heat of fire Wear goggles and self-contained breathing apparatus. DO NOT USE WATER OR FOAM ON FIRE. 4.7 Auto Ignition Temperature: Not pertinent 4.8 Electrical Hazards: Not pertinent CALL FOR MEDICAL AID 4.9 Burning Rate: Not pertinent Exposure 4.10 Adiabatic Flame Temperature: Currently VAPOR not available VAPOR Initiating to eyes, nose and throat. If inhaled will cause coughing or difficult breathing. If in eyes, hold eyelids open and flush with plenty of water. If breathing has stopped, give artificial respiration. If breathing is difficult, give oxygen. 4.11 Stoichometric Air to Fuel Ratio: Not pertinent 4.12 Flame Temperature: Currently not available 4.13 Combustion Molar Ratio (Reactant to Product): Not pertinent Will burn skin and eves. 4.14 Minimum Oxygen Concentration for Combustion (MOCC): Not listed Will ourn skin and eyes. Harmful if swallowed. Remove contaminated clothing and shoes. Flush affected areas with plenty of water. IF IN EYES, hold eyelds open and flush with plenty of water. IF SWALLOWED and victim is CONSCIOUS, have victim drink water or milk. IF SWALLOWED and victim is UNCONSCIOUS OR HAVING CONVULSIONS, do nothing except keep victim warm. 5. CHEMICAL REACTIVITY 5.1 Reactivity with Water: Reacts violently with water, evolving hydrogen fluoride, an extremely irritating and corrosive gas. Reactivity with Common Materials: Reacts violently with many metals and Effect of low concentrations on aquatic life is unknown. May be dangerous if it enters water intakes. Notify local health and wildlife officials. Notify operators of nearby water intakes. 5.2 Water Pollution materials of construction such as wood. glass, some plastics. 5.3 Stability During Transport: Stable 5.4 Neutralizing Agents for Acids and Caustics: Flush with water. 5.5 Polymerization: Not pertinent 1. CORRECTIVE RESPONSE ACTIONS 2. CHEMICAL DESIGNATIONS 2. CHEMICAL DESIGNATIONS CG Compatibility Group: Not listed. Formula: BrFs IMO/UN Designation: 8/1745 DOT ID No: 1745 CAS Registry No.: 7789-30-2 NAERG Guide No.: 144 Standard Industrial Trade Classification: 52241 Dilute and disperse Stop discharge Chemical and Physical Treatment: 5.6 Inhibitor of Polymerization: Not pertinent 21 2.2 2.3 2.4 2.5 6. WATER POLLUTION Neutralize Do not add water to undissolved material 6.1 Aquatic Toxicity: Currently not available Do not burn 2.6 2.7 6.2 Waterfowl Toxicity: Currently not available 6.3 Biological Oxygen Demand (BOD): None 3. HEALTH HAZARDS 6.4 Food Chain Concentration Potential: 3.1 Personal Protective Equipment: Self-contained breathing apparatus, acid suit, and gloves 3.2 Symptoms Following Exposure: Chemical is highly corrosive and toxic. Inhalation causes severe burns of mucous membrane. Ingestion causes severe burns of mouth. Contact with eyes or skin 6.5 GESAMP Hazard Profile Bioaccumulation: 0 Damage to living resources: (3) Human Oral hazard: (3) causes severe burns. 3.3 Treatment of Exposure: Get medical attention IMMEDIATELY for any exposure to this chemical, even if no adverse effects are evident. INHALATION: remove victim from area; apply artificial respiration if breathing has ceased. INGESTION: give large amount of water. EVES: wash with copious amounts of water for 15 min. SKIN: wash with large amounts of water and follow with lime Human Contact hazard: || Reduction of amenities: XX water; remove contaminated clothing. 3.4 TLV-TWA: 0.1 ppm 3.5 TLV-STEL: Not listed. 3.6 TLV-Ceiling: Not listed. 3.7 Toxicity by Ingestion: Currently not available 3.8 Toxicity by Inhalation: Currently not available. 3.9 Chronic Toxicity: Currently not available 3.10 Vapor (Gas) Irritant Characteristics: Currently not available 3.11 Liquid or Solid Characteristics: Currently not available 3.12 Odor Threshold: Currently not available 3.13 IDLH Value: Not listed. 3.14 OSHA PEL-TWA: Not listed. 3.15 OSHA PEL-STEL: Not listed. 3.16 OSHA PEL-Ceiling: Not listed. 3.17 EPA AEGL: Not listed

7. SHIPPING INFORMATION 7.1 Grades of Purity: Technical, 98.0+%; Pure, 99.9% 7.2 Storage Temperature: Ambient 7.3 Inert Atmosphere: Padded 7.4 Venting: Safety relief 7.5 IMO Pollution Category: Currently not available 7.6 Ship Type: Currently not available 7.7 Barge Hull Type: Currently not available 8. HAZARD CLASSIFICATIONS 8.1 49 CFR Category: Oxidizer 8 2 49 CER Class: 5 1 8.3 49 CFR Package Group: | 8.4 Marine Pollutant: No 8.5 NFPA Hazard Classification: Category Classi Health Hazard (Blue)...... Classification Flammability (Red)..... 0 Instability (Yellow)..... 3 Special (White)...... ₩ ох 8.6 EPA Reportable Quantity: Not listed. 8.7 EPA Pollution Category: Not listed. 8.8 RCRA Waste Number: Not listed 8.9 EPA FWPCA List: Not listed 9. PHYSICAL & CHEMICAL PROPERTIES 9.1 Physical State at 15° C and 1 atm: Liquid 9.2 Molecular Weight: 174.9 9.3 Boiling Point at 1 atm: 106°F = 41°C = 314°K 9.4 Freezing Point: -76°F = -60°C = 213°K 9.5 Critical Temperature: 386.6°F = 197°C = 470.2°K 9.6 Critical Pressure: Currently not available 9.7 Specific Gravity: 2.48 at 20°C (liquid) 9.8 Liquid Surface Tension: Currently not 9.9 Liquid Water Interfacial Tension: Not pertinent 9.10 Vapor (Gas) Specific Gravity: 6.03 9.11 Ratio of Specific Heats of Vapor (Gas): 1.089 at 25°C 9.12 Latent Heat of Vaporization: 76.8 Btu/lb = 42.7 cal/g = 1.79 X 10⁵ J/kg 9.13 Heat of Combustion: Not pertinent 9.14 Heat of Decomposition: Not pertinent 9.15 Heat of Solution: Currently not available 9.16 Heat of Polymerization: Not pertinent

9.17 Heat of Fusion: 7.07 cal/g

9.18 Limiting Value: Currently not available

9.19 Reid Vapor Pressure: Currently not available

NOTES

BROMINE PENTAFLUORIDE

9.20 SATURATED LIQUID DENSITY		9.21 LIQUID HEAT CAPACITY		9.22 LIQUID THERMAL CONDUCTIVITY		9.23 LIQUID VISCOSITY	
Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F	Temperature (degrees F)	British thermal unit inch per hour-square foot-F	Temperature (degrees F)	Centipoise
45 50 55 60 65 70 75 80 85 90 95 100 105	157.599 157.000 156.400 155.799 154.599 154.599 153.400 153.400 152.801 152.199 151.599 151.000 150.400		P E R T I N E N T		P Hoursquare root		N O T P E R T I N E N T

9.24 SOLUBILITY IN WATER		9.25 SATURATED VAPOR PRESSURE		9.26 SATURATED VAPOR DENSITY		9.27 IDEAL GAS HEAT CAPACITY	
Temperature (degrees F)	Pounds per 100 pounds of water	Temperature (degrees F)	Pounds per square inch	Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F
	R E A C T S	45 50 55 60 65 70 75 80 85 90 95 100 105 110 110 115 120 125 130 135 140 145 155	3.542 4.039 4.593 5.212 5.899 6.661 7.504 8.436 9.463 10.590 11.330 13.190 14.680 16.300 18.080 20.000 22.100 24.370 26.840 29.500 32.380 35.490 38.830	45 50 55 60 65 70 75 80 85 90 95 100 105 110 110 115 120 125 130 135 140 145 155	0.11430 0.12910 0.14540 0.16340 0.20490 0.22870 0.25470 0.31400 0.34760 0.34760 0.38410 0.42360 0.46630 0.51250 0.56230 0.61590 0.67350 0.73530 0.87250 0.94840 1.02900		N O T P E R T I N E N T