MERCURIC OXIDE

7. SHIPPING INFORMATION 7.1 Grades of Purity: Red-technical; reagent; purified Yellow-technical; NF; reagent

7.5 IMO Pollution Category: Currently not available

8. HAZARD CLASSIFICATIONS

8.5 NFPA Hazard Classification: Not listed 8.6 EPA Reportable Quantity: Not listed.

9. PHYSICAL & CHEMICAL PROPERTIES

9.1 Physical State at 15° C and 1 atm: Solid
9.2 Molecular Weight: 216.61
9.3 Boiling Point at 1 atm: Not pertinent (decomposes)

8.7 EPA Pollution Category: Not listed.
8.8 RCRA Waste Number: Not listed

7.2 Storage Temperature: Ambient

7.4 Venting: Open

7.3 Inert Atmosphere: No requirement

7.6 Ship Type: Currently not available7.7 Barge Hull Type: Currently not available

8.1 49 CFR Category: Poison 8.2 49 CFR Class: 6.1

8.3 49 CFR Package Group: ||

8.9 EPA FWPCA List: Not listed

9.4 Freezing Point: Not pertinent

pertinent

Not pertinent

NOTES

9.5 Critical Temperature: Not pertinent9.6 Critical Pressure: Not pertinent

9.7 Specific Gravity: 11.1 at 20°C (solid)

9.8 Liquid Surface Tension: Not pertinent 9.9 Liquid Water Interfacial Tension: Not

9.10 Vapor (Gas) Specific Gravity: Not pertinent

9.11 Ratio of Specific Heats of Vapor (Gas):

9.14 Heat of Decomposition: Not pertinent9.15 Heat of Solution: Not pertinent

9.16 Heat of Polymerization: Not pertinent

9.17 Heat of Fusion: Currently not available
9.18 Limiting Value: Currently not available
9.19 Reid Vapor Pressure: Currently not available

9.12 Latent Heat of Vaporization: Not pertinent 9.13 Heat of Combustion: Not pertinent

8.4 Marine Pollutant: Yes

CAUTIO	NARY RESPO	ONSE INFORMATION		4. FIRE HAZARDS		
Common Synonyms Mercuric oxide, red Mercuric oxide, yellow Mercury oxide	Solid Sinks in water.	Red, orange or yellow Odorless	4	 Flash Point: Not flammable, but may intensify fire Flammable Limits in Air: Not flammable Fire Extinguishing Agents: Not pertinent 		
KEEP PEOPLE AWAY. Wear dust respirator ar Notify local health and p	AVOID CONTACT W ad rubber overclothing pollution control agenci	ITH SOLID AND DUST. (including gloves). es.	4	Special Hazards of Combustion Products: Fumes from fire may contain		
Fire Not flamm Will increa POISONC	able. Ise the intensity of a fi US GASES MAY BE F	re. RODUCED WHEN HEATED.	4	 Behavior in Fire: Decomposes at 500°C into mercury and oxygen, which can increase intensity of fire. Solid changes 		
Exposure CALL FOR DUST POISONO If inhaled If in eyes, If breathin POISONO Initiating to If swallow Remove c Flush aff IF IN EYE IF SWALL	CALL FOR MEDICAL AID. DUST POISONOUS IF INHALED OR IF SKIN IS EXPOSED. If inhaled will cause coughing or difficult breathing. If in eyes, hold eyelids open and flush with plenty of water. If breathing has stopped, give artificial respiration. If breathing is difficult, give oxygen. SOLID POISONOUS IF SWALLOWED OR IF SKIN IS EXPOSED. Irritating to skin and eyes. If swallowed will cause nausea or vomiting. Remove contarninated clothing and shoes. Flush affected areas with plenty of water. IF IN EYES, hold eyelids open and flush with plenty of water. IF SWALLOWED and victim is CONSCIOUS, have victim drink water or mik and have victim induce vomiting. IF SWALLOWED and victim is UNCONSCIOUS OR HAVING CONVULSIONS, do nothing except keep victim warm. Effect of low concentrations on aquatic life is unknown. May be dangerous if it enters water intakes. Notify local heath and wildlife officials. Notify operators of nearby water intakes.			 4.7 Auto Ignition Temperature: Not pertinent 4.8 Electrical Hazards: Not pertinent 4.9 Burning Rate: Not pertinent 4.10 Adiabatic Flame Temperature: Curren not available 4.11 Stoichometric Air to Fuel Ratio: Not pertinent. 4.12 Flame Temperature: Currently not available 4.13 Combustion Molar Ratio (Reactant to Product): Not pertinent. 4.14 Minimum Oxygen Concentration for Combustion (MOCC): Not listed 5. CHEMICAL REACTIVITY 5.1 Reactivity with Water: No reaction 5.2 Reactivity with Common Materials: Currently not available 5.3 Stability During Transport: Stable 5.4 Neutralizing Agents for Acids and Caustics: Not pertinent 5.5 Polymerization: Not pertinent 		
Water Pollution Notify ope						
1. CORRECTIVE RESPONS Stop discharge Collection Systems: Dr	SE ACTIONS edge	2. CHEMICAL DESIGNATION 2.1 CG Compatibility Group: Not listed 2.2 Formula: HgO 2.3 IMO/UN Designation: 6.1/1641 2.4 DOT ID No.: 1641 2.5 CAS Registry No.: 1344-45-2 2.6 NAERG Guide No.: 151 2.7 Standard Industrial Trade Classif 52269	S d. 6 ication: 6	Inhibitor of Polymerization: Not pertinent 6. WATER POLLUTION Aquatic Toxicity: 0.29 ppm/48 hr/marine fish/TLm Xvaterfowl Toxicity: Currently not available Biological Oxygen Demand (BOD): None 4. Food Chain Concentration Potential:		
 Personal Protective Equip by uremic poisoning 14 concentrations of 1.2-8 and difficulty in breathin with eyes causes ulcer possible dermatitis; sys is whites, milk, or activate for at least 15 min. SKI 4 TLV-TWA: 0.025 mg/m² (add) 5 TLV-STEL: Not listed. TLV-TWA: 0.025 mg/m² (add) 5 TLV-STEL: Not listed. TLV-Ceiling: Not listed. TLV-Ceiling: Not listed. Toxicity by Inhalation: Cur. Chronic Toxicity: Causes 1 10 Vapor (Gas) Irritant Characteri 12 Odor Threshold: Odorless 13 IDLH Value: Not listed. OSHA PEL-TWA: Not listed. OSHA PEL-TWA: Not listed. OSHA PEL-TWA: Not listed. OSHA PEL-Ceiling: 0.1 mg 17 EPA AEGL: Not listed 	 S. HEALITH S. HEALITH Tensk: gog Sugari Acute systemic issually delayed 5-12 di 5 S mg/m² of ari, sympti S mg/m² of ari, sympti S mg/m² of ari, sympti Tenski acuto Tenski acuto	gles or face shield; protective gloves poisoning may be fatal within a few minutes ays. Acute poisoning has resulted from inhal poisoning may be fatal within a few minutes ays. Acute poisoning has resulted from inhal cornea. Contact with skin causes irritation cour by absorption through skin. icitim to fresh air; get medical attention. INGf ing first 10-15 min. determines prognosis. G vomiting; consult physician. EYES; flush wit d water. g/kg (rat) ot available ailable	: death 6 Iing dust Jing, Contact a and SSTION: ive egg h water	mercury from water. Bioconcentrative up to 10,000 fold. 55 GESAMP Hazard Profile: Bioaccumulation: + Damage to living resources: 4 Human Oral hazard: 3 Human Contact hazard: 1 Reduction of amenities: XXX		

JUNE 1999

MERCURIC OXIDE

9.20 SATURATED LIQUID DENSITY		9.21 LIQUID HEAT CAPACITY		9.22 LIQUID THERMAL CONDUCTIVITY		9.23 LIQUID VISCOSITY	
Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F	Temperature (degrees F)	British thermal unit inch per hour-square foot-F	Temperature (degrees F)	Centipoise
	N O T		N O T		N O T		N O T
	- PERT-NENT		I P E R T I N E N T		- PERT-NENT		- PERT-NENT

9.24 SOLUBILITY IN WATER		9.25 SATURATED VAPOR PRESSURE		9.26 SATURATED VAPOR DENSITY		9.27 IDEAL GAS HEAT CAPACITY	
Temperature (degrees F)	Pounds per 100 pounds of water	Temperature (degrees F)	Pounds per square inch	Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F
	- z % 0		N O T		N O T		N O T
	O L U B L E		P E R T I N E N T		P E R T I N E Z T		P E R T I N E N T