VANADIUM OXYTRICHLORIDE

CAUTIONARY RESPONSE INFORMATION Common Synonyms Sharp unpleasant Trichlorooxovanadium Vanadyl chloride Vanadyl trichloride Sinks and mixes violently with water KEEP PEOPLE AWAY. AVOID CONTACT WITH LIQUID AND VAPOR Wear rubber overclothing (including gloves). Notify local health and pollution control agencies Do not add water to undissolved chemical. Fire DO NOT USE WATER ON ADJACENT FIRES. CALL FOR MEDICAL AID. **Exposure** VAPOR If in eyes, hold eyelids open and flush with plenty of water. If breathing has stopped, give artificial respiration. If breathing is difficult, give oxygen. LIQUID POISONOUS IF SWALLOWED OR IF SKIN IS EXPOSED. Irritating to skin and eyes. Harmful if swallowed. Harmful if swallowed. Remove contaminated clothing and shoes. Flush affected areas with plenty of water. IF IN EYES, hold eyelids open and flush with plenty of water. IF SWALLOWED and victim is CONSCIOUS, have victim drink water or milk. IF SWALLOWED and victim is UNCONSCIOUS OR HAVING CONVULSIONS, do nothing except keep victim warm. Effect of low concentrations on aquatic life is unknown. Water May be dangerous if it enters water intakes **Pollution** Notify local health and wildlife officials. Notify operators of nearby water intakes

1. CORRECTIVE RESPONSE ACTIONS

Dilute and disperse dissolved mate Stop discharge Chemical and Physical Treatment:

Do not add water to undissolved material

2. CHEMICAL DESIGNATIONS

- CG Compatibility Group: Not listed.
- Formula: VOCla

- IMO/UN Designation: Not listed DOT ID No.: 2443 CAS Registry No.: 7727-18-6 NAERG Guide No.: 137
 - Standard Industrial Trade Classification: 52329

3. HEALTH HAZARDS

- 3.1 Personal Protective Equipment: Acid vapor mask; rubber gloves; face shield; acid-resistant clothing
- 3.2 Symptoms Following Exposure: Inhalation of vapor causes irritation of nose and throat. Ingestion causes irritation of mouth and stomach. Contact with eyes or skin causes severe irritation.
- 3.3 Treatment of Exposure: Consult a physician after all exposures to this compound. INHALATION: move to fresh air; give artificial respiration if necessary. INGESTION: give large amount of water. EYES: flush with water for 15 min. StNN: wipe exposed areas free of the chemical with a dry cleb the of fine the complex. cloth, then flush thoroughly with water.
- 3.4 TLV-TWA: Not listed.
- 3.5 TLV-STEL: Not listed
- 3.6 TLV-Ceiling: Not listed.
- 3.7 Toxicity by Ingestion: Grade 3; oral rat LD₅₀ = 140 mg/kg
- 3.8 Toxicity by Inhalation: Currently not available.
- 3.9 Chronic Toxicity: Repeated exposures may cause discoloration of tongue, loss of appetite, anemia, kidney disorders, and blindness.
- 3.10 Vapor (Gas) Irritant Characteristics: Currently not available 5.9 Currently not available 5.10 10 ppm HCl, based on decomposition of compound in moist air
- 3.11 Liquid or Solid Characteristics: Currently not available 3.12 Odor Threshold: Currently not available
- 3.13 IDLH Value: Not listed. 3.14 OSHA PEL-TWA: Not listed.
- 3 15 OSHA PEL-STEL: Not listed
- 3.16 OSHA PEL-Ceiling: Not listed.
- 3.17 EPA AEGL: Not listed

4. FIRE HAZARDS

- Flash Point: Not flammable
- 4.2 Flammable Limits in Air: Not flammable
- 4.3 Fire Extinguishing Agents: Not pertinent
- 4.4 Fire Extinguishing Agents Not to Be
 Used: Water, unless in flooding amounts, should not be used on adjacent fires.
- 4.5 Special Hazards of Combustion Products: Irritating fumes of hydrogen chloride may form in fires.
- 4.6 Behavior in Fire: Currently not available
- 4.7 Auto Ignition Temperature: Not pertinent
- 4.8 Electrical Hazards: Not pertinent
- 4.9 Burning Rate: Not pertinent
- 4.10 Adiabatic Flame Temperature: Currently
- 4.11 Stoichometric Air to Fuel Ratio: Not
- 4.12 Flame Temperature: Currently not available
- 4.13 Combustion Molar Ratio (Reactant to Product): Not pertinent.
- 4.14 Minimum Oxygen Concentration for Combustion (MOCC): Not listed

5. CHEMICAL REACTIVITY

- **5.1 Reactivity with Water:** Reacts to form a solution of hydrochloric acid
- Reactivity with Common Materials: In presence of moisture will corrode most
- 5.3 Stability During Transport: Stable
- 5.4 Neutralizing Agents for Acids and Caustics: Flush with water and sprinkle with powdered limestone or rinse with dilute solution of sodium bicarbonate or soda ash.
- 5.5 Polymerization: Not pertinent
- 5.6 Inhibitor of Polymerization: Not pertinent

6. WATER POLLUTION

- 6.1 Aquatic Toxicity: See vanadyl sulfate
- 6.2 Waterfowl Toxicity: Currently not 6.3 Biological Oxygen Demand (BOD): None
- 6.4 Food Chain Concentration Potential:
- Currently not available
- 6.5 GESAMP Hazard Profile: Bioaccumulation: 0 Damage to living resources: (2) Human Oral hazard: 2

Human Contact hazard: II Reduction of amenities: XX

7. SHIPPING INFORMATION

- 7.1 Grades of Purity: Technical, 99%
- 7.2 Storage Temperature: Ambient
- 7.3 Inert Atmosphere: No requirement
- 7.4 Venting: Pressure-vacuum
- 7.5 IMO Pollution Category: Currently not available
- 7.6 Ship Type: Currently not available
- 7.7 Barge Hull Type: Currently not available

8. HAZARD CLASSIFICATIONS

- 8.1 49 CFR Category: Corrosive material
- 8.2 49 CFR Class: 8
- 8.3 49 CFR Package Group: II
- 8.4 Marine Pollutant: No
- 8.5 NFPA Hazard Classification:

Category Classification Health Hazard (Blue)........ 3 Flammability (Red)..... Instability (Yellow).....

- 8.6 EPA Reportable Quantity: Not listed.
- 8.7 EPA Pollution Category: Not listed.
- 8.8 RCRA Waste Number: Not listed
- 8.9 EPA FWPCA List: Not listed

9. PHYSICAL & CHEMICAL PROPERTIES

- 9.1 Physical State at 15° C and 1 atm: Liquid
- 9.2 Molecular Weight: 173.3
- 9.3 Boiling Point at 1 atm: 259°F = 126°C =
- 9.4 Freezing Point: -107°F = -77°C = 196°K
- 9.5 Critical Temperature: Currently not available
- 9.6 Critical Pressure: Currently not available
- 9.7 Specific Gravity: 1.83 at 20°C (liquid)
- 9.8 Liquid Surface Tension: Currently not
- 9.9 Liquid Water Interfacial Tension: Not pertinent
- 9.10 Vapor (Gas) Specific Gravity: 5.98
- 9.11 Ratio of Specific Heats of Vapor (Gas): Currently not available
- 9.12 Latent Heat of Vaporization: Currently not
- 9.13 Heat of Combustion: Not pertinent 9.14 Heat of Decomposition: Not pertinent
- 9.15 Heat of Solution: Currently not available
- 9.16 Heat of Polymerization: Not pertinent 9.17 Heat of Fusion: Currently not available
- 9.18 Limiting Value: Currently not available
- 9.19 Reid Vapor Pressure: Currently not available

NOTES

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9.20 SATURATED LIQUID DENSITY		9.21 LIQUID HEAT CAPACITY		9.22 LIQUID THERMAL CONDUCTIVITY		9.23 LIQUID VISCOSITY	
Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F	Temperature (degrees F)	British thermal unit inch per hour-square foot-F	Temperature (degrees F)	Centipoise
35 40 45 50 55 60 65 70 75 80 85 90 95	116.500 116.200 115.590 115.599 115.299 115.000 114.700 114.400 114.000 113.700 113.400 113.099 112.799 112.500	0 5 10 15 20 25 33 40 45 50 55 60 65 70 75 80 85 90 95	0.125 0.125		NOT PERTINENT		NOT PERTINENT

9.24 SOLUBILITY IN WATER		9.25 SATURATED VAPOR PRESSURE		9.26 SATURATED VAPOR DENSITY		9.27 IDEAL GAS HEAT CAPACITY	
Temperature (degrees F)	Pounds per 100 pounds of water	Temperature (degrees F)	Pounds per square inch	Temperature (degrees F)	Pounds per cubic foot	Temperature (degrees F)	British thermal unit per pound-F
	REACTS	60 70 80 90 100 110 120 130 140 150 160 170 180 190 200 210 220 230 240 250 260	0.190 0.256 0.340 0.448 0.584 0.755 0.967 1.228 1.547 1.933 2.400 2.958 3.623 4.409 5.335 6.417 7.678 9.139 10.820 12.760 14.970	60 70 80 90 100 110 120 130 140 150 160 170 180 200 210 220 230 240 250 260	0.00591 0.00780 0.01018 0.01316 0.01316 0.01686 0.02140 0.02693 0.03361 0.04163 0.05120 0.06252 0.07585 0.09144 0.10960 0.13060 0.15470 0.18240 0.21390 0.24970 0.29020 0.33580		NOT PERTINENT